

# **Revision Sheet** The Primary Stage of Grades (4-5)

2023-2024

Name:	Subject: Science
Date: / /	<b>Class:</b> Grade 4 (C, D, E, F, G)

#### Question 1:

a)	Defi	ine the rate of evaporation.
b)	Stat	e the factors that affect the evaporation rate:
	1.	
	2.	

### **Question 2:**

Anastasia wants to measure the temperature of water in the cooking pot before putting it over the heat. What does she use to accurately measure the temperature? Circle the correct answer and state its name.

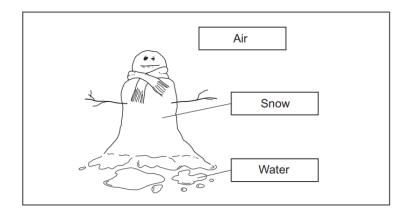






#### **Question 3:**

Sarah has built a snowman on a cold but sunny day.



a)	Identify	v the	states	of	matter	of	the	fol	lowing	?
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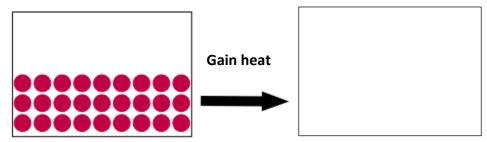
Air:	Snow:	Water :

b) State the process that is happening to the snowman in the sun?

Describe what happens to the particles during this process.

.....

c) Describe in drawing the change that happens to the particles of ice.



d) After several hours water started to evaporate.

What name is given to water in the gaseous state? .....

#### **Question 4:**

In the diagram below, arrows **J,K,L** and **O** represent processes involving the changes in states of matter.



a) State the name of the process that each arrow represents:

J:..... K:.....

L:......

b) In process K, how can a gas can turn back into a liquid?

Describe what happens to the particles during this change.

.....

## **Question 5:**

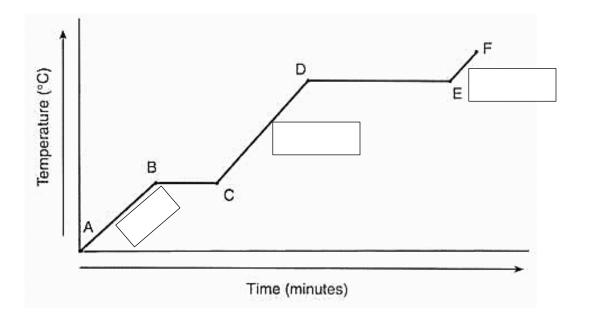
name of process

Boiling, freezing and melting are three processes that water can go through. Draw a line to match each process to what happens to the water during the process.

what happens during the process

	boiling		ice gains heat and changes into a liquid		
	freezing		water gains heat and changes into a gas		
	melting		water loses heat and changes into a solid		
Quest	ion 6:				
Anast	asia is investigat	ting the melting	and boiling points of pure water.		
She p	She puts some ice in the cooking pot over a flame and heats it.				
The ic	e starts to melt.	,			
a)	What will happ	en to the <b>liquid</b>	water as it gains heat?		
 لم	Λ ft or οbilo		hail Haw on Anastasia know that the		
b) After a while, water started to boil. How can Anastasia know that the					
water is boiling?					
c)	State the name	given to <b>the te</b>	mperature at which a liquid starts to boil?		

d) Anastasia represented her results in the following line graph:



#### Study the graph and answer the following:

- 1. Which point represents the melting point of ice? ......
- 2. Which point represents the boiling point of water? .....
- 3. On the graph, decide the area where water is in the solid state, liquid state and gaseous state.
- 4. What happens to the temperature in the area **B-C**?

5. What happens to the temperature in the area **D-E**?

.....

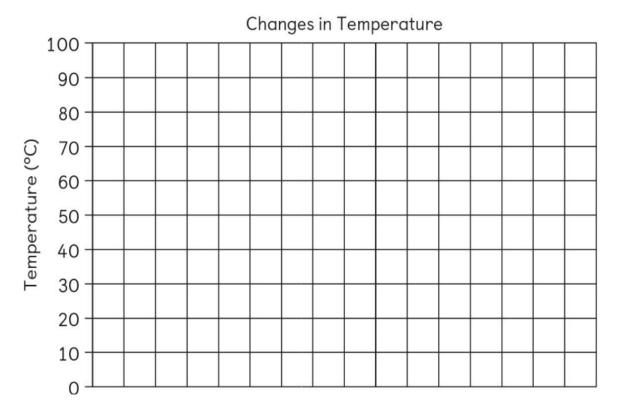
- 6. At which temperature does pure water boil. .....
- 7. State the melting point of pure water. .....

# **Question 7:**

Anastasia measures the temperature every two minutes. She recorded her results in the table below:

Time (Minutes)	Temperature (°C)
0	20
2	40
4	50
6	70
8	90
10	100
12	100
14	100

### Draw a line graph that shows her results



## **Question 8:**

a.	Sand	owing substance Jelly Powder	Clay	Iron
	Suga	ar Salt	Flour	
	Soluk	ole	In	soluble
In th	- Sugar:		sugar:	
C.	•			tain the sea salts dissolved i
	this water. Desc	cribe the method	d used to obtai	n these salts.

#### **Question 9:**

After **3 days**, they n

Maya and Rami are planning an investigation to determine if the surface area of the container affects the rate of evaporation.

They poured 100 ml of water in each of the beakers shown below and left it **for 3 days** at the room temperature.

ne amount of water that was left.

Beaker A Beaker B **1.** Identify the following in this investigation: Dependent variable: Independent variable: ..... Control variable: ..... 2. Predict, from which beaker will water evaporate faster? **3.** Complete the following conclusion: The ...... the surface area of a container, the ..... the rate of evaporation is.

## Question 10:

Tick (V) the correct box beside each statement:

Statement	Solid	Liquid	Gas
Has a fixed shape and volume			
Particles can be compressed			
Particles are tightly packed			
Takes the shape of the container			
Particles spread randomly			
Particles vibrate in their place			
Particles slide against each other			
Has no fixed shape or volume			

