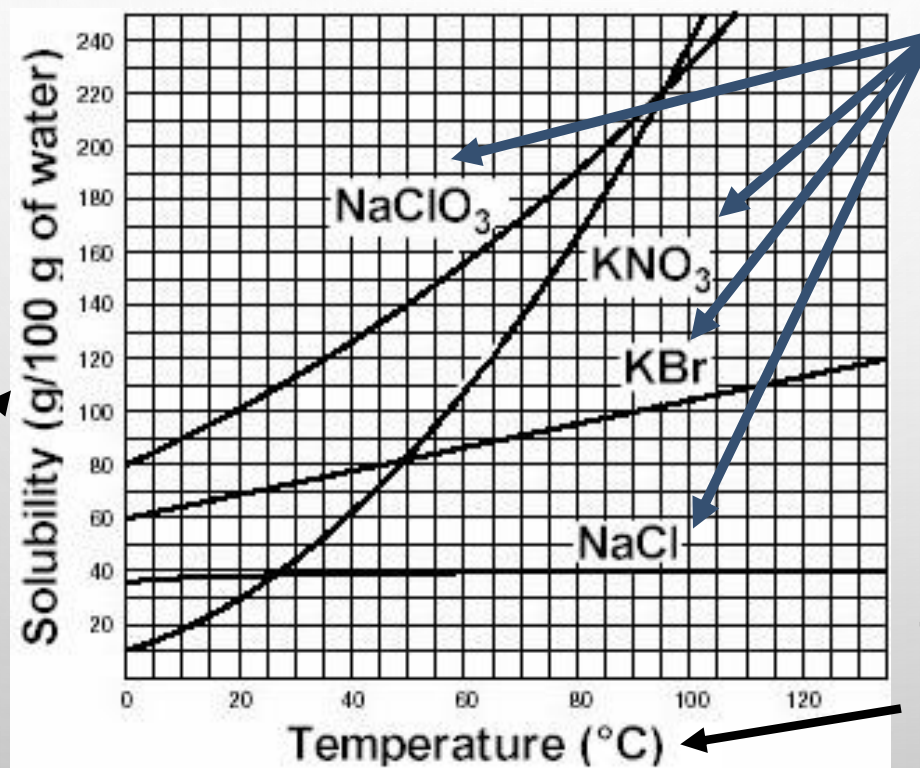


# SOLUBILITY CURVES

HOW TO INTERPRET A GRAPHICAL REPRESENTATION OF SOLUTE IN SOLVENT.

# SOLUBILITY CURVE

- **SOLUBILITY CURVE** – A GRAPHICAL REPRESENTATION OF THE AMOUNT OF SUBSTANCE THAT CAN DISSOLVE INTO 100 G OF WATER AT A SPECIFIC TEMPERATURE (CELSIUS)



## Substances:

Compound being dissolved in water (H<sub>2</sub>O)

## Y-axis:

Solubility of substance (g/100 g H<sub>2</sub>O)

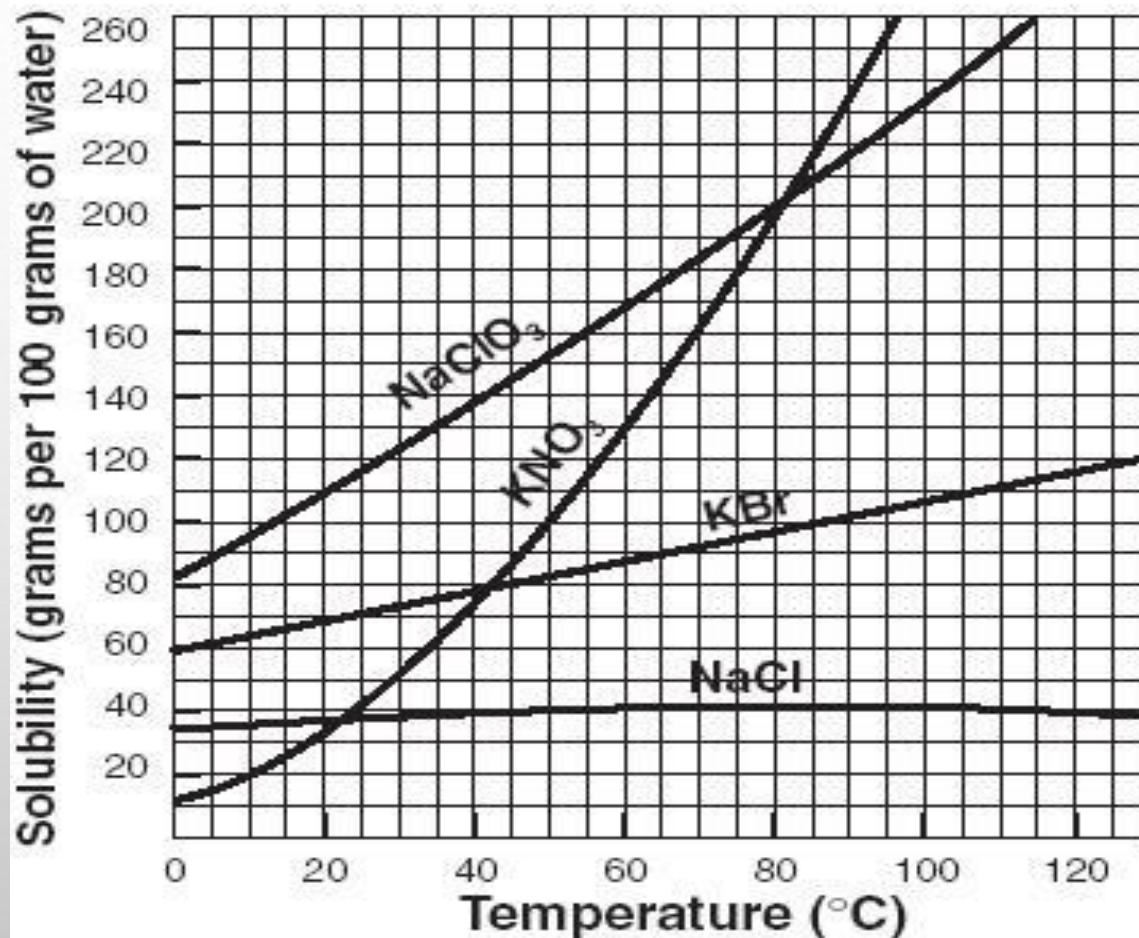
## X-axis:

Temperature (Celsius)

# INTERPRETING A SOLUBILITY CURVE

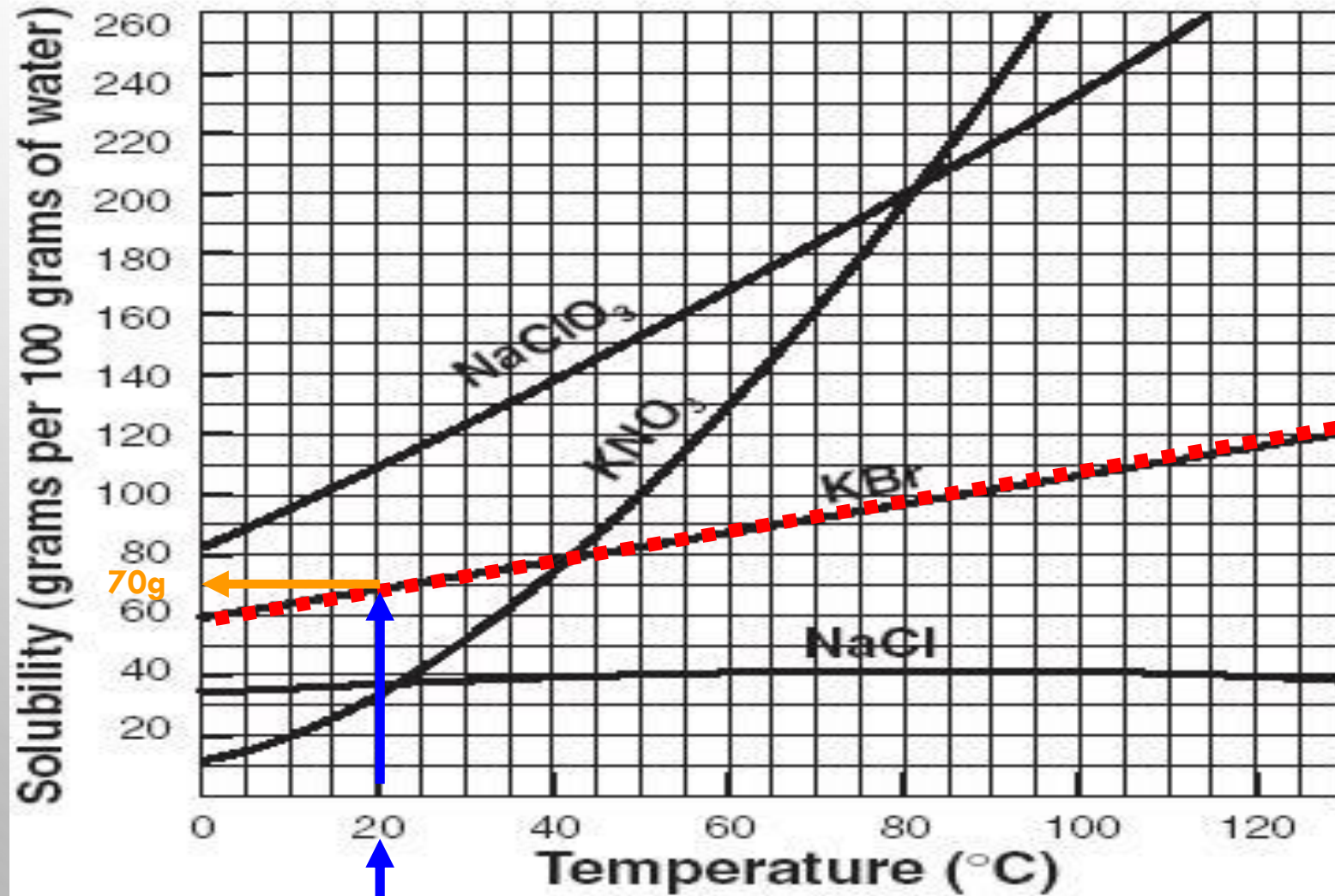
- EACH POINT ON THE SOLUBILITY CURVE SHOWS HOW MANY GRAMS CAN BE DISSOLVED AT A SPECIFIC TEMPERATURE:

Each line shows how much substance can dissolve as a **function** of the temperature of the solution.



# USING A SOLUBILITY CURVE

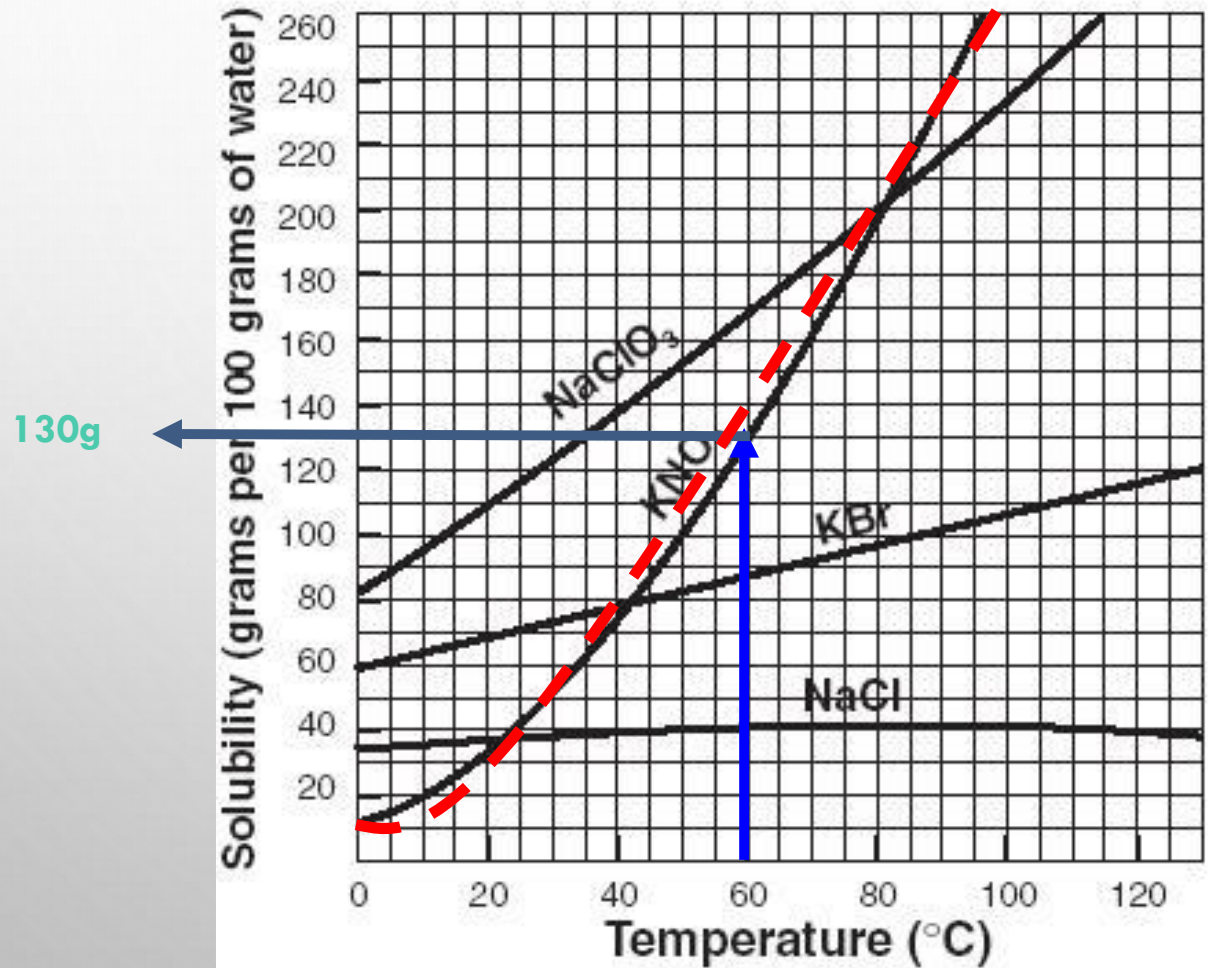
How many grams of potassium bromide (KBr) can dissolve in 100 grams of water at 20°C?



**Answer:** 70 grams of KBr can dissolve in 100g of water at 20°C

# PRACTICE USING SOLUBILITY CURVE

How many grams of potassium nitrate ( $\text{KNO}_3$ ) can dissolve in 100 g of water at  $60^\circ\text{C}$ ?

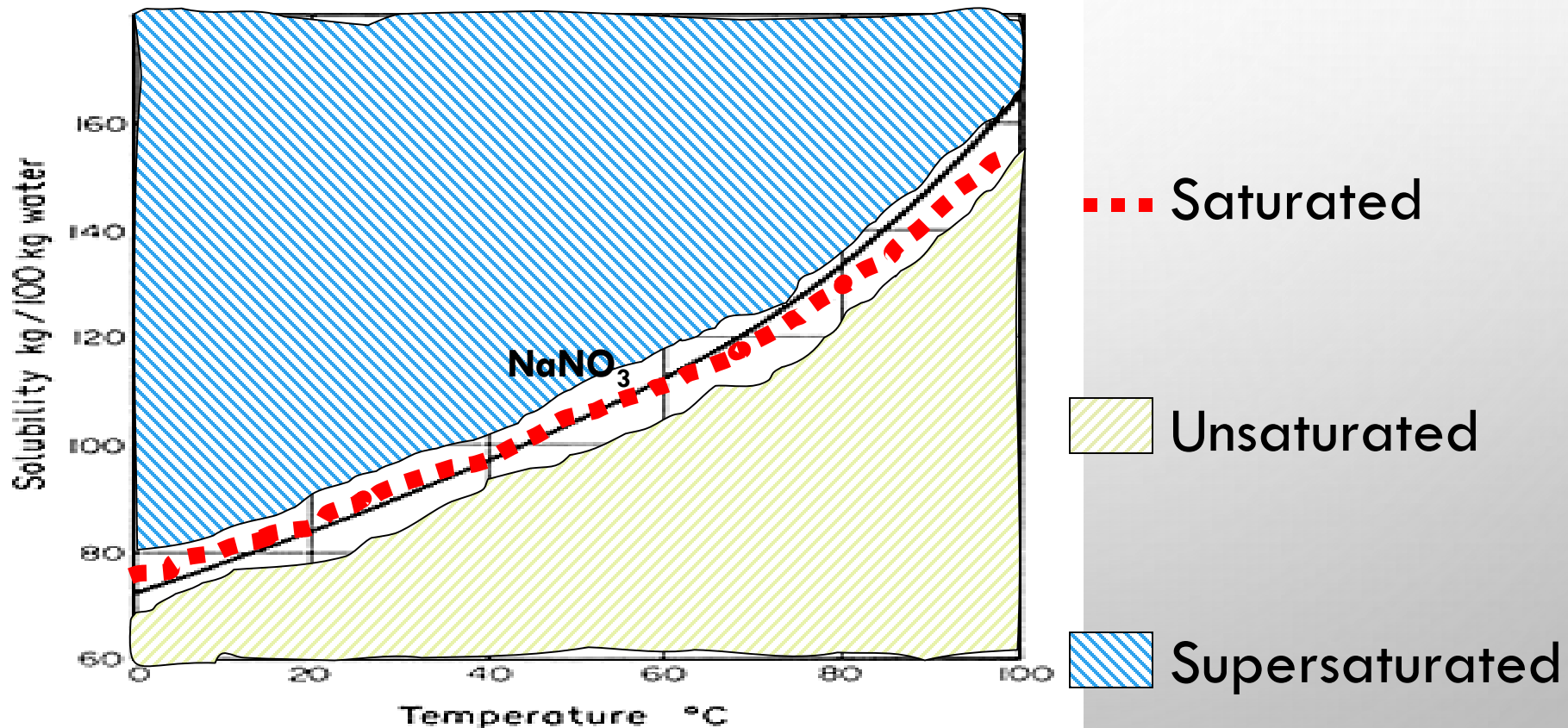


**Answer:** 130 g of  $\text{KNO}_3$  can dissolve in 100 g of  $\text{H}_2\text{O}$

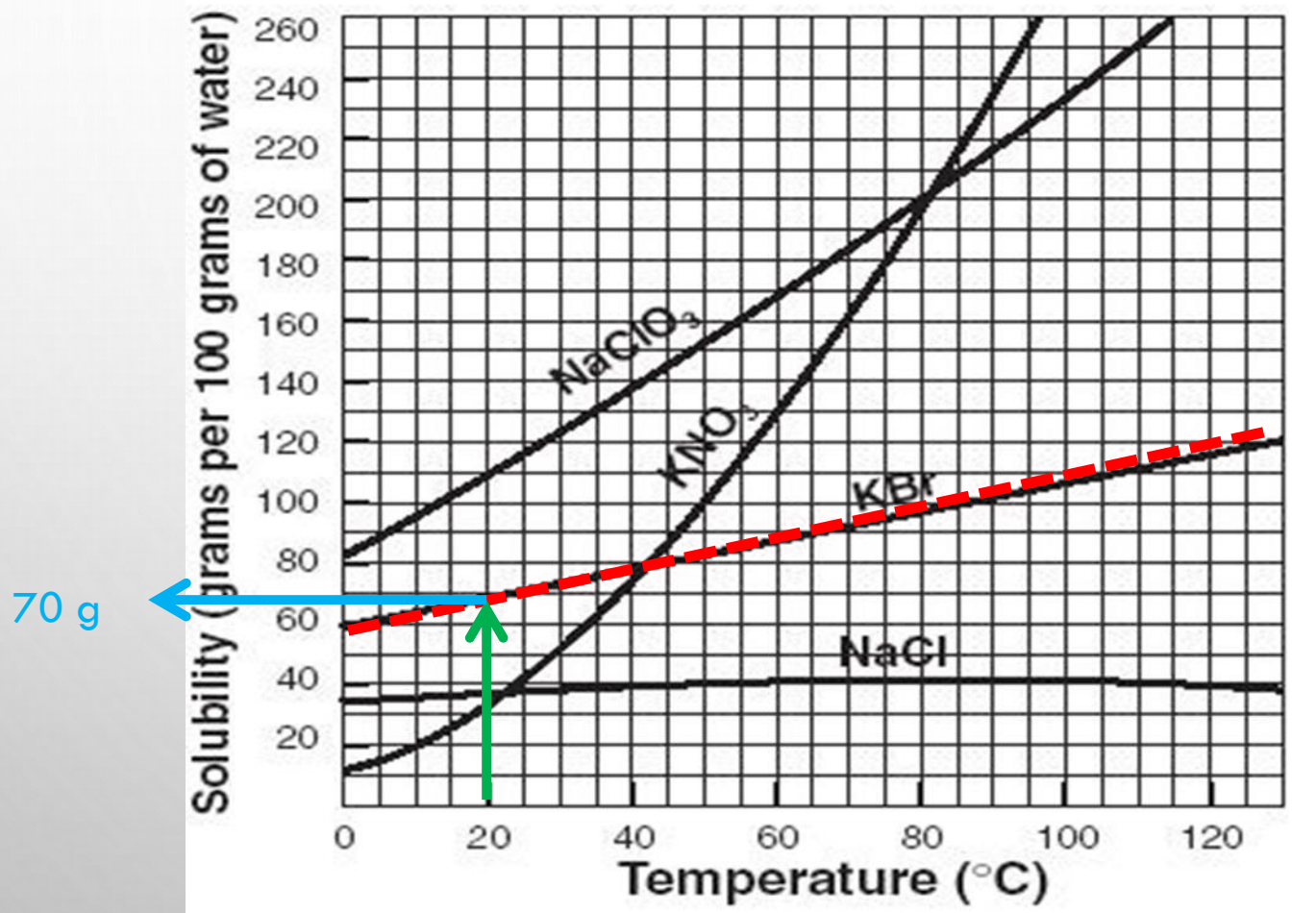
# SATURATED / UNSATURATED / SUPERSATURATED

**REVIEW:** HOW WE NUMERICALLY DESCRIBE SATURATION

- SATURATED: SOLUTE = SOLUBILITY
- UNSATURATED: SOLUTE < SOLUBILITY
- SUPERSATURATED: SOLUTE > SOLUBILITY

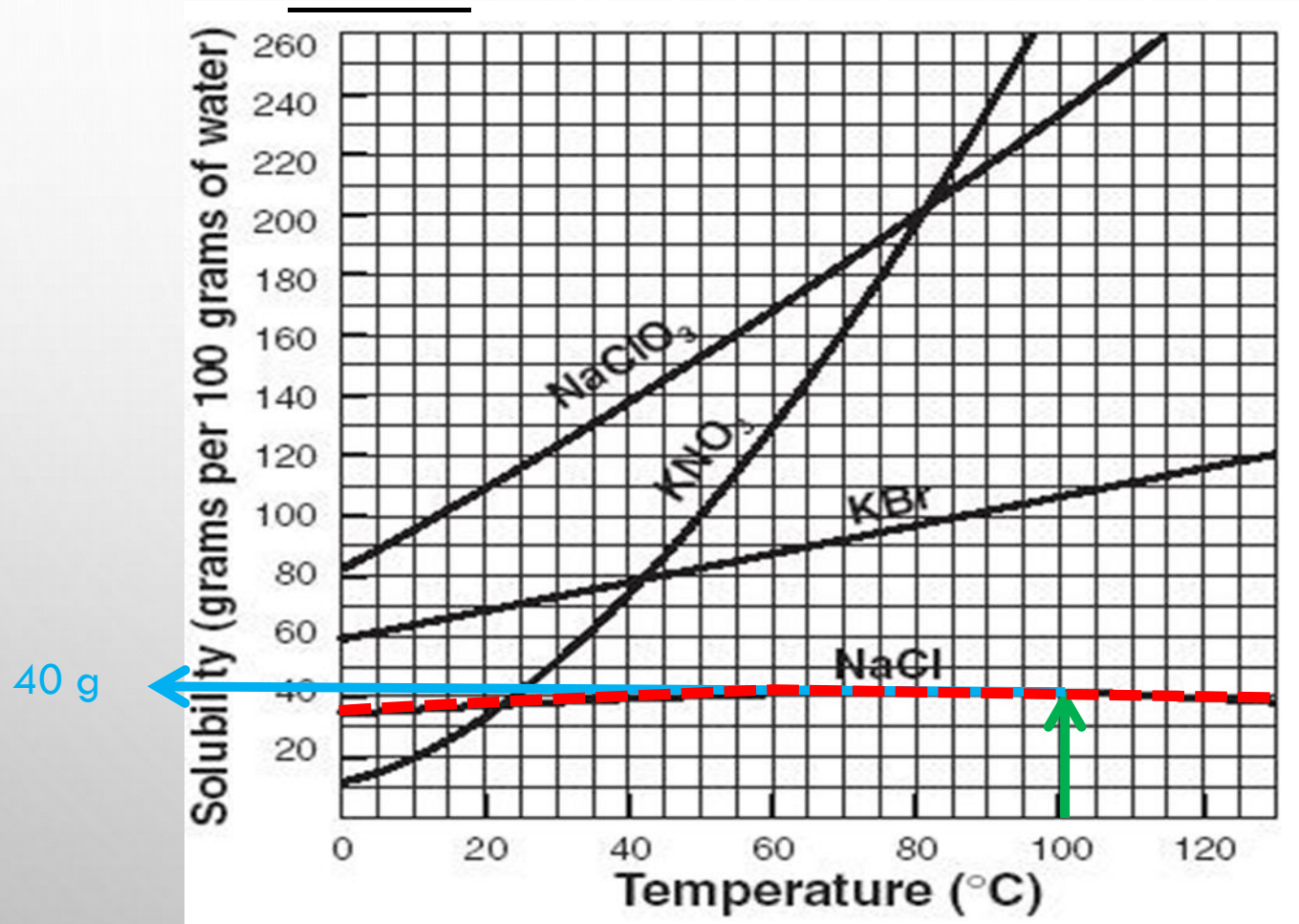


Practice #1: How many grams of potassium bromide (KBr) can dissolve in 100 g of water at 20°C?



Answer: 70 grams of Potassium Bromide can be dissolved in 100 grams of water at 20° C.

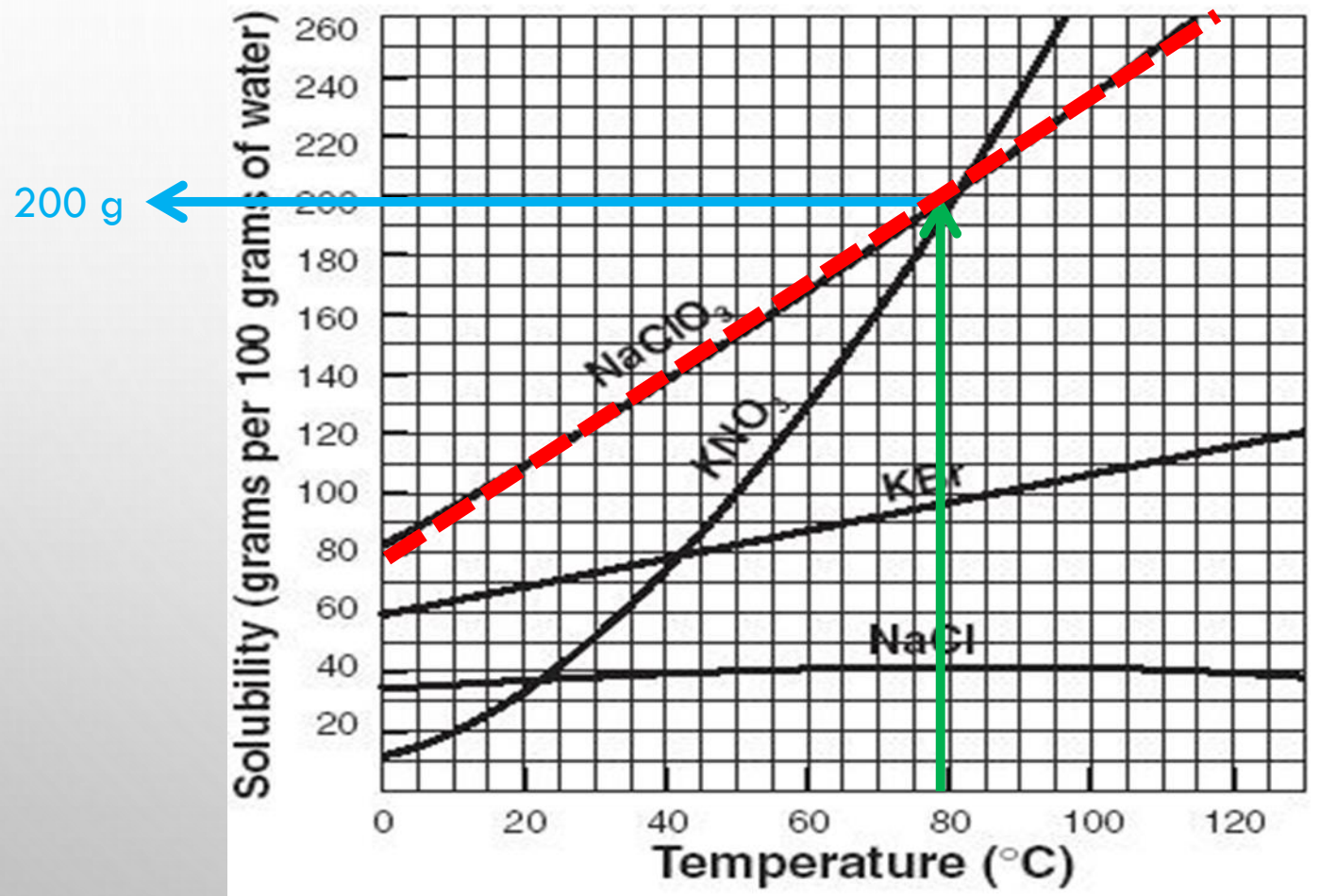
Practice #2: How many grams of sodium chloride (NaCl) can dissolve in 100 g of water at 100°C?



Answer: 40 grams of Sodium chloride can be dissolved in 100 g of water at 100°C



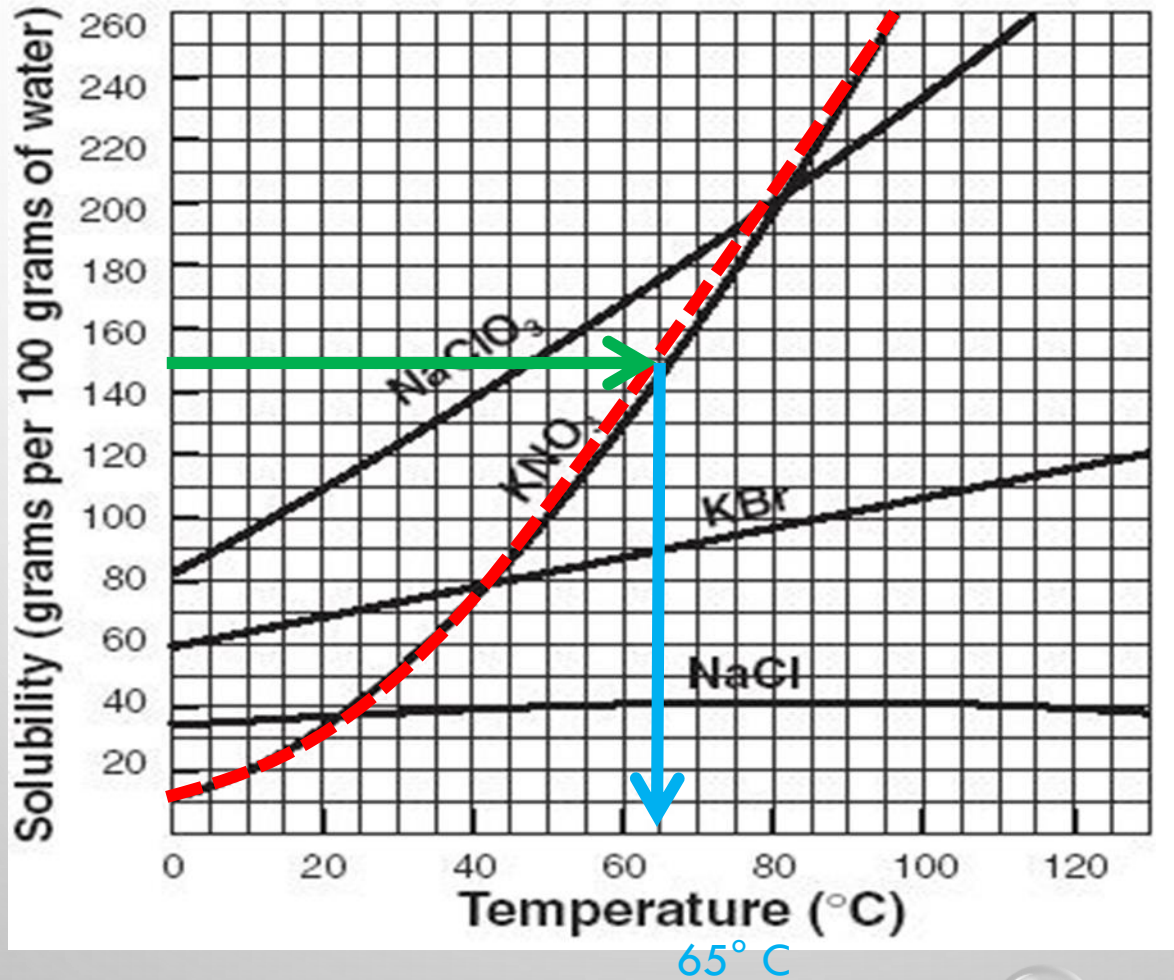
Practice #3: How many grams of sodium chlorate ( $\text{NaClO}_3$ ) can dissolve in 200 g of water at  $80^\circ\text{C}$ ? \_\_\_\_\_



200g per 100 g of water, so in 200 g of water we will have to double it:

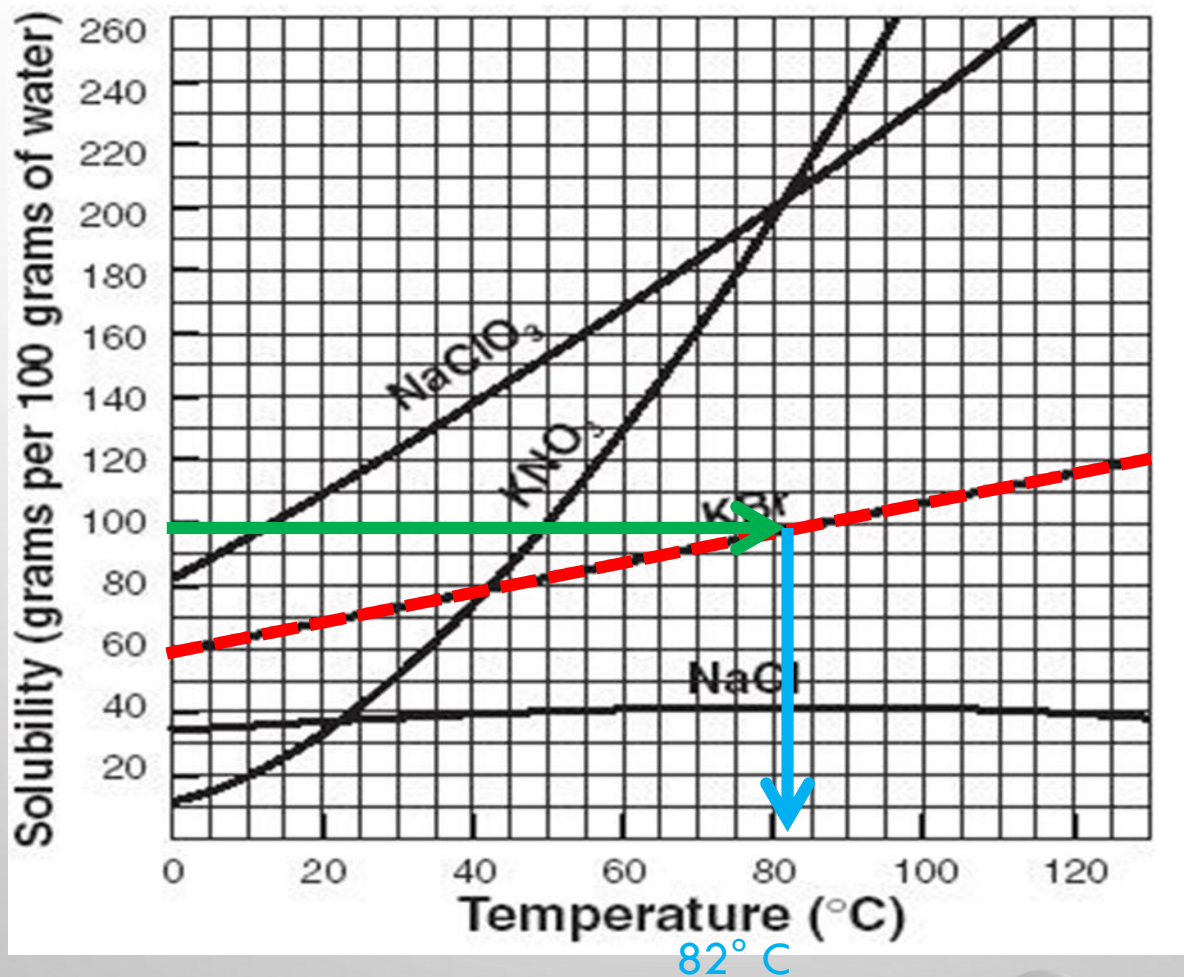
$200 \times 2 = 400$  g  $\text{NaClO}_3$  can be dissolved in 200 g of water at  $80^\circ\text{C}$

Practice #4: At what temperature can 150 grams of potassium nitrate ( $\text{KNO}_3$ ) dissolve in 100 g of water? \_\_\_\_\_



Answer: 150 grams of Potassium nitrate can be dissolved in 100 g of water at 65°C

Practice #5: At what temperature can 100 grams of potassium bromide (KBr) dissolve in 100 g of water?



Answer: 100 g of potassium bromide can dissolve in 100 g of water at 82°C