



The National Orthodox School/ Shmessani

Name:

Worksheet: Solubility curve

Date:

Grade 7CS -ALL Sections

Objective: To be able to read solubility curves.

1. Using the solubility curve, find the solubility for each salt:

- KNO_3 at 50°C
- NaNO_3 at 25°C
- $\text{Ce}_2(\text{SO}_4)_3$ at 20°C

2. A student dissolves 120g of NaNO_3 at 60°C , his solution is

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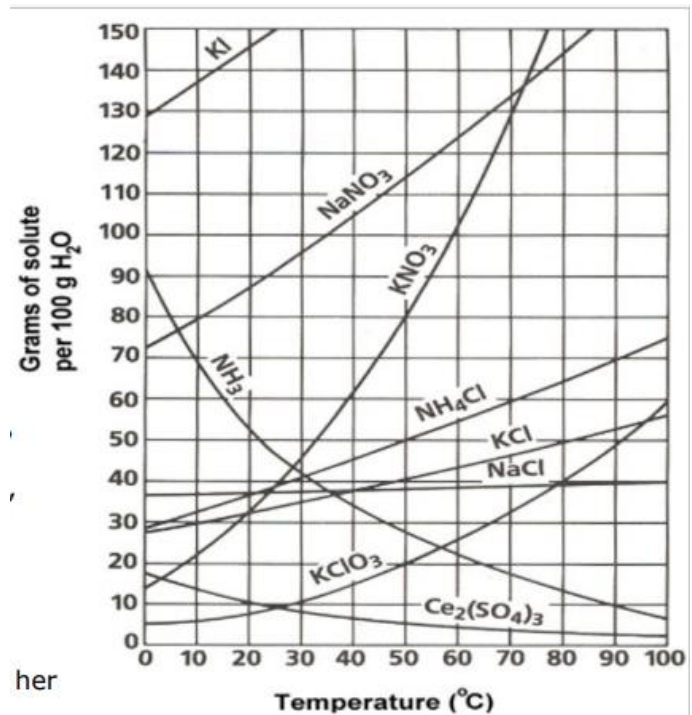
3. A student dissolves 80g of KCl at 90°C , her solution is

4. A student dissolves 129g of KI at 0°C , his solution is

.....

5. Which compound is the least soluble?

6. Which compound is the most soluble?





1. Using the solubility curve, find the solubility for each salt:
 - KNO_3 at 50°C **80g**
 - NaNO_3 at 25°C **90g**
 - $\text{Ce}_2(\text{SO}_4)_3$ at 20°C **10g**
2. A student dissolves 120g of NaNO_3 at 60°C , his solution is **...unsaturated..**
3. A student dissolves 80g of KCl at 90°C , her solution is **...super saturated...**
4. A student dissolves 129g of KI at 0°C , his solution is **..saturated..**
5. Which compound is the least soluble? **.. $\text{Ce}_2(\text{SO}_4)_3$..**
6. Which compound is the most soluble? **.. KI ..**

