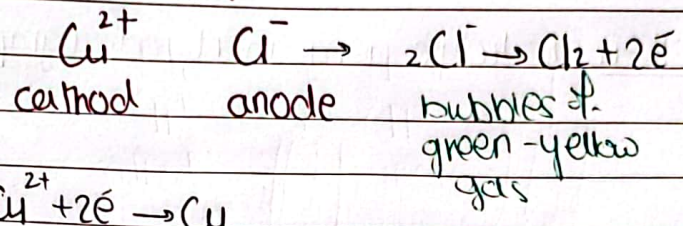
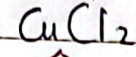
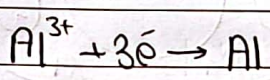
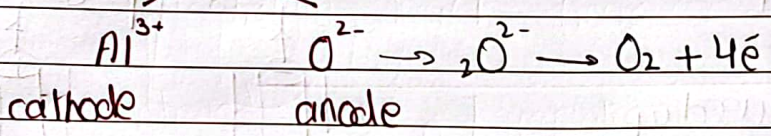
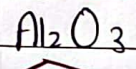


1)



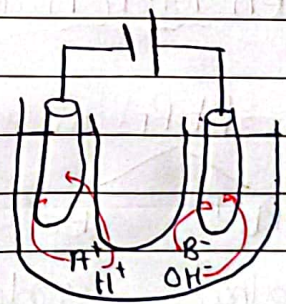
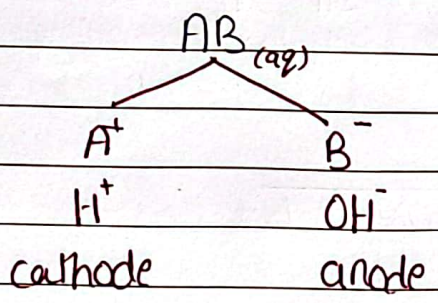
$Cu^{2+} + 2e^- \rightarrow Cu$
 Deposit of red-brown solid

Electrolyte: consume



Electrolyte: used up.

Electrolysis for aqueous electrolyte using Inert.



* At the cathode: - less reactive more likely to reduce and the more reactive stays in the electrolyte.