

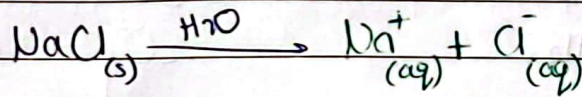
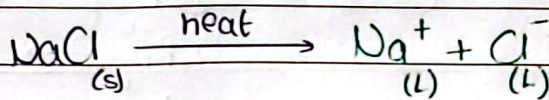
Electrolysis

Electricity      Analysis  
(Breaking down)

\* Electrolysis: breaking down chemical ~~reaction~~ compounds (ionic) when molten or aqueous by passing electricity.

\* Why the ionic compound don't conduct electricity when solid?  
 → the ions are not free to move.

\* Why the ionic compounds conduct electricity when dissolve in water or being molten?  
 → the ions are free to move.



\* Electrolyte: chemical compound that conduct electricity when molten or aqueous.

