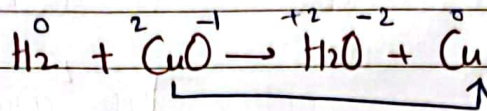
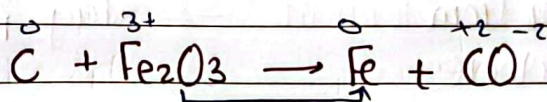


* Most common reducing agent:-

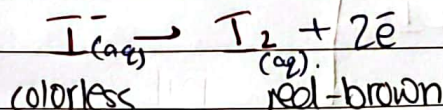
(1) Hydrogen.



(2) Carbon and carbon monoxide



(3) Iodide



(4) Metals

- Metals more reactive, more likely to lose e-
- more likely to oxidise.
- " " to be a reducing agent.

K → strongest reducing agent.
Ag → weakest " "

Ag⁺ → strongest oxidising agent.
K⁺ → weakest " "