

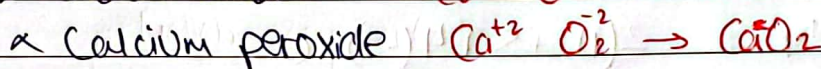
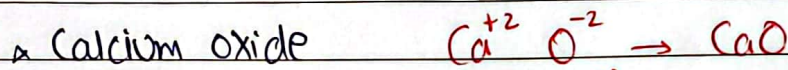
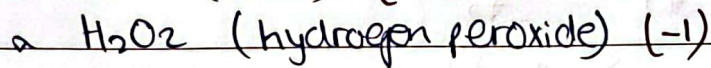
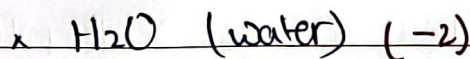
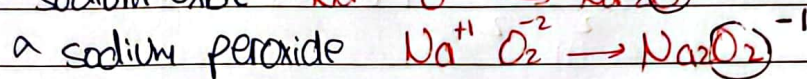
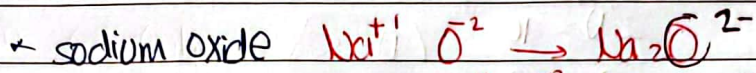
4 The oxidation state of oxygen (-2)  
 except in peroxide (-1)  
 " "  $F^n OF_2$  (2+)



oxide



peroxide



5- The sum of all oxidation state in a compound = 0  
 in an ion = charge of this ion.

