Starters

The solid state:

* The particles touch each other.
* The particles are in fixed positions in a regular pattern.
* The particles vibrate on the spot.
* The particles hold very strongly together.

 The liquid state:

* The particles touch each other.
* They are not in a pattern.
* The particles hold strongly.
* The particles move around randomly sliding over each other.

 The gas state:

* The particles are far apart.
* They are not in a pattern.
* The particles move around very fast in all directions.
* The particles do not hold together strongly



Main course

|  |  |  |  |
| --- | --- | --- | --- |
| Process | Change of state | Temperature  | How particles leave the liquid |
| Evaporating | From liquid to gas | Happens at any temperature  | Particles leave the liquid surface  |
| Boiling  | From liquid to gas | Happens at the boiling point  | Bubbles of the substance in the gas state form liquid; the bubbles rise to the surface and escape  |

Dessert

